

Recitation Nuts and Bolts

Agi Stachowiak

Instructor, 20.109 and 20.300
(contributions to 20.110, 20.111, and 20.309)

August 26th, 2014

Goals of a recitation

- Reinforce lecture material
 - stick to essentials
 - reframe, synthesize
- Teach problem-solving skills
 - demos *and* practice!
 - make steps and assumptions transparent
- Provide meaningful opportunities for interaction
 - not just: “any questions?”
 - one option: “think, pair, share” exercises
- More from Rotem this afternoon!

Anatomy of a recitation

- Typical breakdown
 - $\frac{1}{4}$ to $\frac{1}{3}$ lecture (max $\frac{1}{2}$)
 - $\frac{2}{3}$ to $\frac{3}{4}$ problems (min $\frac{1}{2}$)
- Types of problems
 - algorithmic
 - usually similar to pset or exam
 - teach robust, transferrable strategies
 - conceptual
 - may or may not show up directly on exams
 - regardless, will help in novel applications

Writing on the board

- Board writing order
 - middle board first → stays visible when...
 - ... top board next → stays visible when...
 - ... bottom board last (if at all; don't write too low).
- Practice! Check from the *back* of the room.
 - big enough?
 - legible?
- Staying within the board neatly
 - see how much content you normally fit on a board
 - map it to a custom-sized index card

Marked up index card examples

- Next few examples come from 20.110 F09
- Layout logistics
- Pedagogical/content tips

Outline for that session

write on board *before* class starts

Announcements

upcoming exams, OH, etc.

20.110 Recitation 3

1.

agenda

- enthalpy: concepts + math $\rightarrow$ H vs. U, expansions
- adiabatic vs. isothermal expansions $\rightarrow$ problem
- TD cycles $\rightarrow$ problem
- index card: something that remains confusing

next time: thermochemistry; CP and protein folding

- * WRITE REC. # ON NEXT HW
- * PROBLEMS PLITS: 1+2, 3+4, 5+6, 7+8
- * OH tonight 6-7 10-536

what is enthalpy (H)?

analogue to U (molec. PE + KE) $U = U(T, V)$ $H = H(T, P)$

(x, y) are "natural variables" of $z(x, y) \rightarrow$ useful in TD, SM

$H \equiv U + PV$ * state function *

macro 1st law	macro if "Y" constant	diff. if "Y" constant	ideal gas
U: $\Delta U = Q + W$	$\Delta U_V = Q_V$	$(dU)_V = C_V dT$ * or $n C_V dT$	$(dU)_{any} = C_V dT$
H: $\Delta H = \Delta U + \Delta(PV)$	$\Delta H_P = \Delta U + P\Delta V$ $\therefore \Delta H_P = Q_P - P\Delta V + P\Delta V$	$(dH)_P = C_P dT$ * or $n C_P dT$	$(dH)_{any} = C_P dT$

Lecture review: synthesize, systematize

Board height markings

Board width marking

Intro to stat mech: multiplicities

$$W = \frac{N!}{n_1! n_2! \dots n_t!}$$

N = total objects

t = distinguishable categories of objects

 n_i = # of (indistinguishable) objects in i^{th} categoryprobability = $\frac{\text{part}}{\text{whole}}$

Short lecture segment followed
by immediate application
may repeat 2-3x per recitation

D&B # 1-2

Note where problems come from

D&B = textbook

a.) AAA TCG AAA

whole = 4^9

part = 1

$$P_a = \frac{3.8 \cdot 10^6}{4^9}$$

N = 9 $n_A = 4$ $n_T = 2$ $n_G = 2$ $n_C = 1$

$$\text{part} = W = \frac{9!}{4! 2! 2! 1!} = 3780$$

b.) AAAAAAAAAA $P_b = P_a$

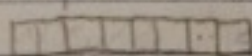
$$P_c = 3780 / 4^9 = 0.014$$

D&B # 2-1

lattice model of gas

15 particles on 20 or

16, or 15 sites - W's?



• are indistinguishable

a.) N = 20 sites categories: + or - •

$$W = \frac{N!}{n_+! n_-!} = \frac{20!}{15! 5!} = 15,504$$

b.) N = 16, $n_+ = 15$, $n_- = 1 \rightarrow 16$ c.) N = 15, $n_+ = 15$, $n_- = 0 \rightarrow 15$

Problems for Recitation 14 (11.02.09, 9 am session)

1. Features of cooperative transitions.

We talked about two features of cooperative transitions that you can see graphically.

- (a) Out loud, explain/discuss how a sigmoidal curve indicates cooperativity.
- (b) What are x and y for the 2-state helix-coil model? Be careful and specific with your definition of y .
- (c) Out loud, explain/discuss how the change in steepness of said sigmoidal curve with system size is in the expected direction. You might use the 2-state model as an example for your discussion.

2. Arbitrary partition functions for helix-coil models.

In class, we discussed examples of non-cooperative and fully cooperative helix-coil systems with independent residues. Using a similar type of partition function as in those cases, answer the following:

- a) Only partial helix-coil states can exist. The molecule may never be fully coiled or fully helical. What is $q(s)$ for a 4-mer? Does this model seem physically likely?
- b) Unlike a protein with a defined N and C terminus, this molecule looks the same going forwards or backwards through its chemical structure. What is $q(s)$ for a 4-mer? Does this model seem physically likely? What is $\theta(s, N)$?

Recitation number and date

Conceptual/discussion problem

Algorithmic problem(s)

Hand out at the beginning so students can work at own pace, during lecture part if they want to.

Summary and related topics

- Recitations
 - a mix of lecture and problems
 - emphasis on problem-solving approaches
 - broader than just homework content
- Review sessions
 - like a super-recitation
 - both lecture *and* problems are still valuable
- Office hours
 - usually for addressing homework questions